

## Reactor type for CaO looping cycle – Bubbling bed or fast bed, that is the question –



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## Reaction in the present process

### CO<sub>2</sub> removal from flue gas by CaO

$\text{CaO} + \text{CO}_2 \rightarrow \text{CaCO}_3$  at 873K ( $P_{\text{CO}_2} = 0.004$  atm)

Exothermic reaction → Steam generation

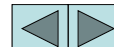
### Regeneration of CaO from CaCO<sub>3</sub>

$\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$  at 1223K ( $P_{\text{CO}_2} = 1.9$  atm)

Endothermic reaction → Combustion of fuel by O<sub>2</sub>

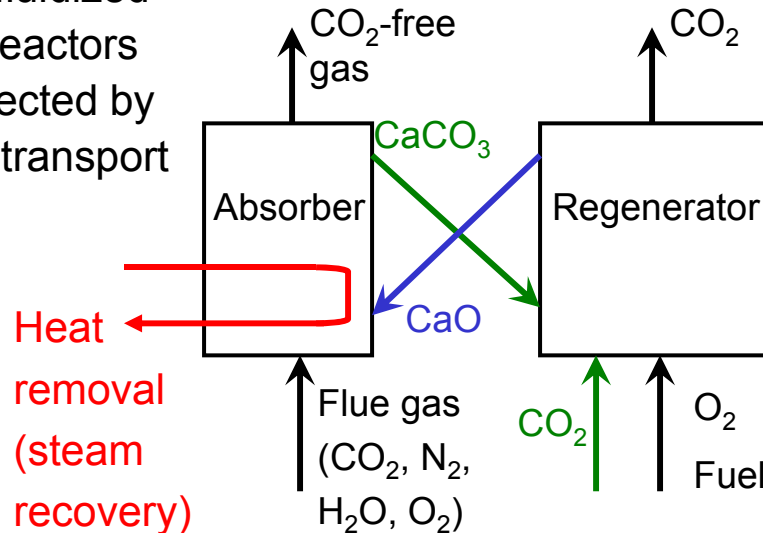
CaO and CaCO<sub>3</sub>: solids

Two fluidized bed reactors connected by solid transport lines.



## Carbonate looping cycle

Two fluidized bed reactors connected by solid transport lines.



## Possible combination

Now we are planning to make an experimental apparatus of this process.

Both bubbling fluidized bed and “fast” fluidized bed are available for reactors. Four possible combinations are:

Carbonation reactor	Regenerator(Calciner)
Bubbling	Bubbling
Bubbling	Fast
Fast	Bubbling
Fast	Fast

## Importance of heat removal

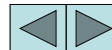
CO<sub>2</sub> removal from flue gas by CaO

$\text{CaO} + \text{CO}_2 \rightarrow \text{CaCO}_3$  at 873K ( $P_{\text{CO}_2} = 0.004 \text{ atm}$ )

Exothermic reaction

→ Heat removal is necessary to maintain reactor temperature.

Often reactor size is determined not by reaction rate but by arrangement of heat transfer surface.



Heat removal scheme and  
required heat transfer surface  
area

## Present system

Based on basic design by present authors  
(Shimizu et al., Trans. Inst. Chem. Eng. , 77, 62, 1999).

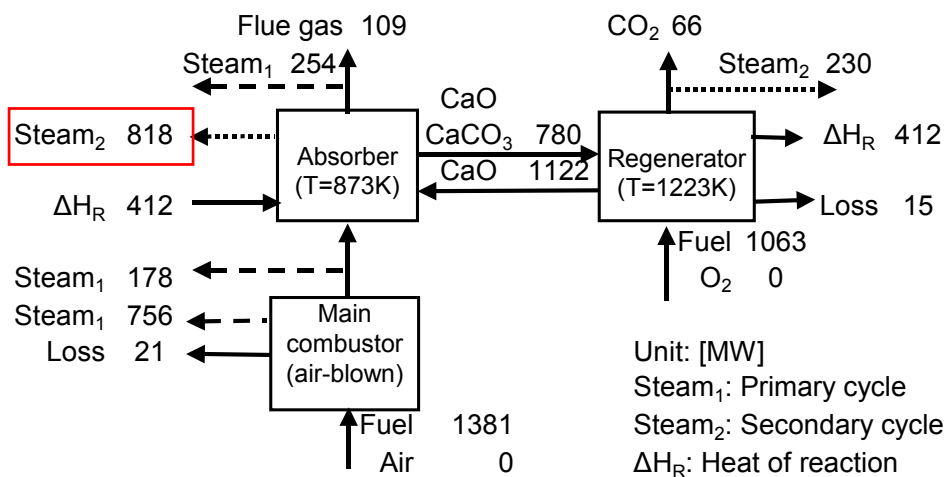
1000 MWe gross output (original concept)  
→ 350 MWe gross (this work)

CaO circulation / captured CO<sub>2</sub> = 10

Subcritical steam from CO<sub>2</sub> capture process  
(170 atm, 566°C, with reheater)

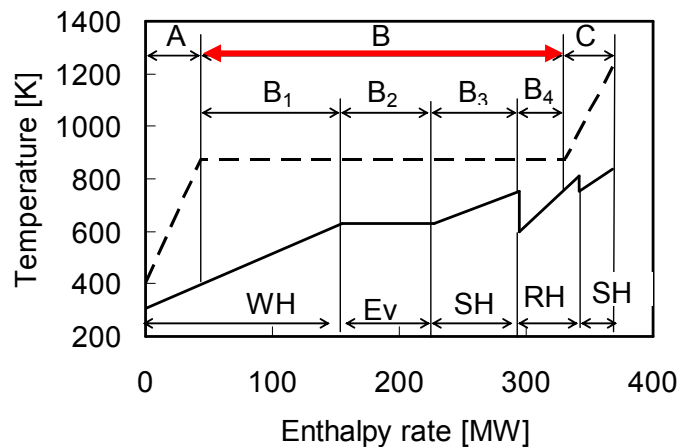
## Heat balance for 1000 MWe(gross) system

Heat removal requirement is very severe.



## Steam recovery from CO<sub>2</sub> capture

Heat source: Carbonation reactor (B) and hot CO<sub>2</sub> from regenerator (A and C). Estimation of heat transfer surface area is conducted.



## Heat transfer surface design

	Bubbling	Fast
Heat transfer surface	tube	flat panel
Bed-surface HT coeff.	258 W/m <sup>2</sup> K	250 W/m <sup>2</sup> K
Gas velocity	1.38 m/s	6 m/s

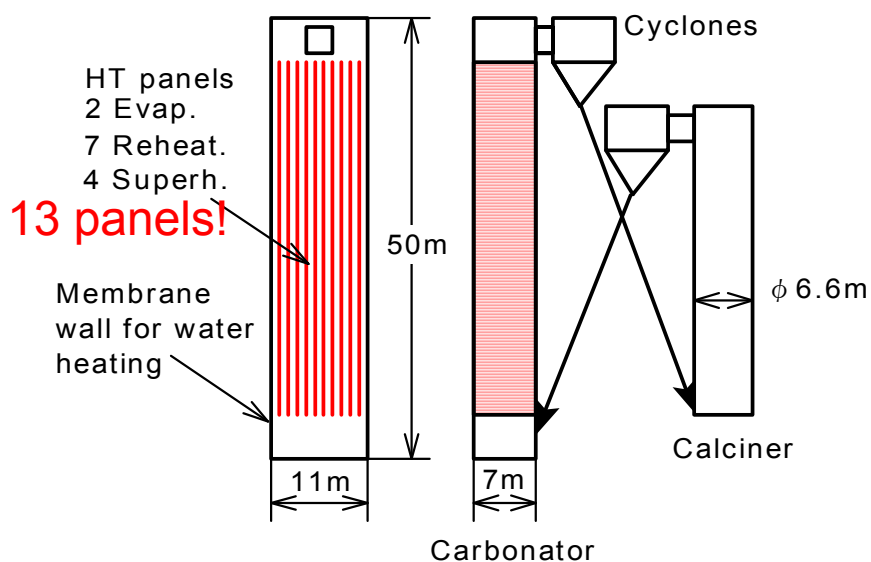
From heat transfer rate, heat transfer coefficient, and temperature difference, required heat transfer surface area of each section was calculated.

## RESULTS AND DISCUSSION

Reactor design to satisfy heat removal requirement.

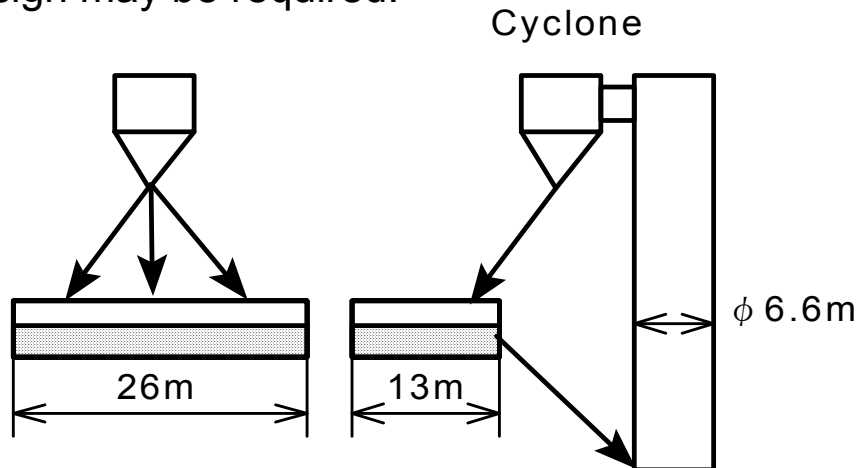
### Fast-fluidized bed carbonator

13(!) HT panels in 11m-width are required.



## Bubbling-fluidized bed carbonator

Bed height is 2.5 m + jet region (0.3 m). Compact but needs more cross-sectional area. Stacked design may be required.



## To reduce heat transfer area in carbonation reactor

Higher reaction temperature

→ Limitation of equilibrium ( $< 923\text{ K}$ )

Lower steam temperature

→ Decreased turbine efficiency

Simple steam cycle without reheating

→ Problem with turbine

Integration of steam cycle with air combustor

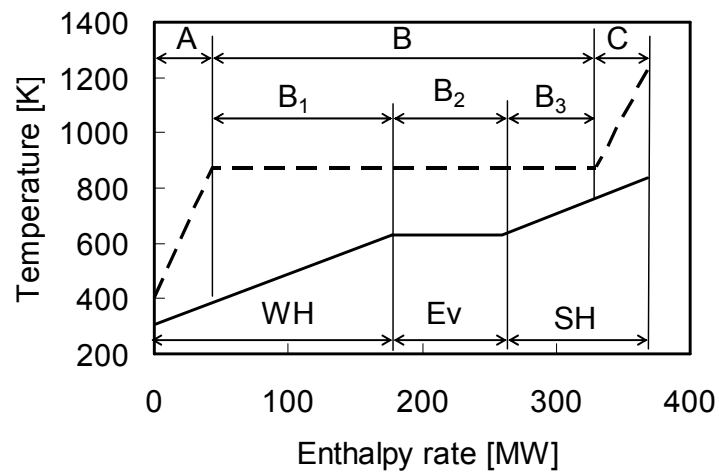
→ Increased construction cost

## Steam recovery from CO<sub>2</sub> capture

Simple superheating without reheating.

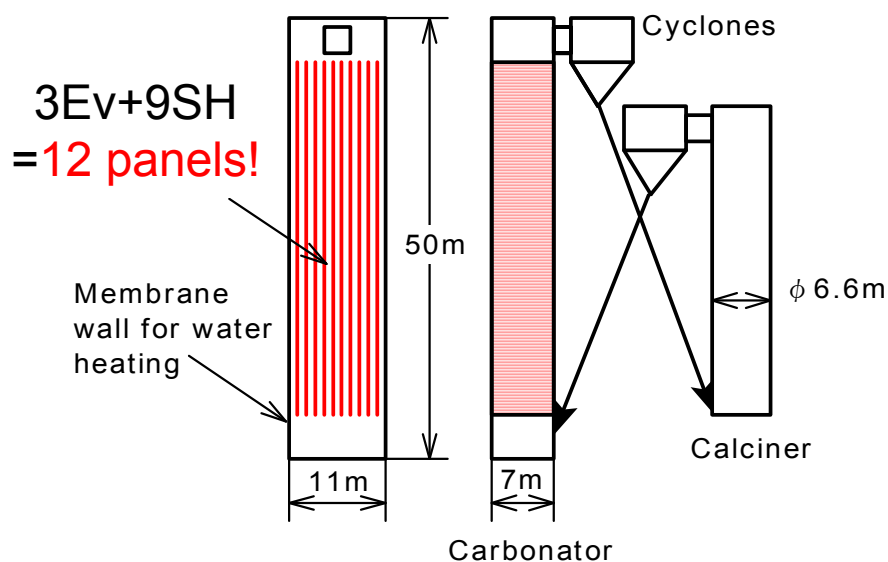
→ More heat recovery by WH and EV.

→



## Fast-fluidized bed carbonator

12(!) HT panels in 11m-width are required.





## RESULTS AND DISCUSSION

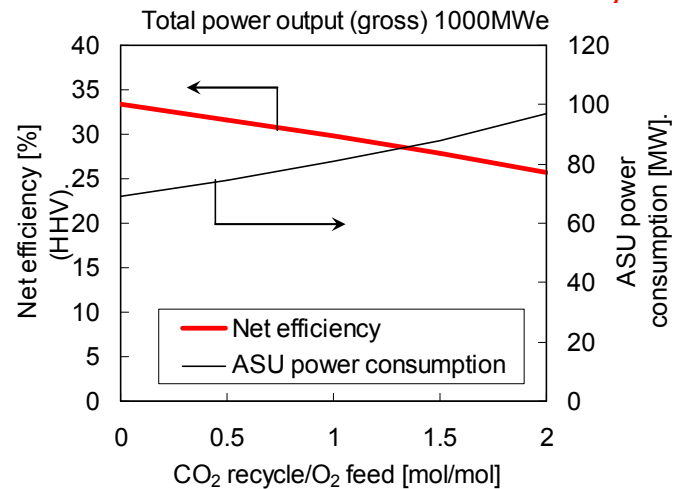
Reactor design to minimize  $O_2$  consumption.

### Regenerator (calcliner)

Combustion in  $O_2/CO_2$  atmosphere

→ Prevention of hot-spot

→ Increased heat loss from the calciner by  $CO_2$



## Regenerator (calciner)

To suppress hot-spot formation under high  $O_2$  partial pressure conditions,

→ "fast" bed may be advantageous, but

→ **loss of sorbent by attrition may become a problem.** (Synthetic sorbent may be a solution though the cost will be higher than natural limestone.)

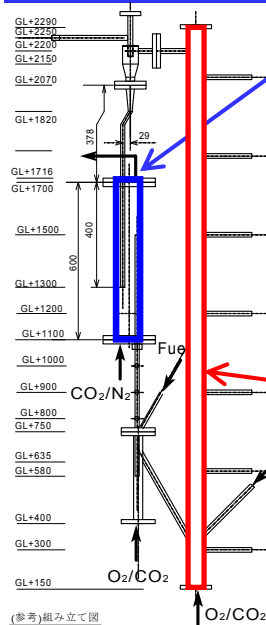
## Conclusion

To remove heat from carbonation reactor, quite large heat transfer surface area is required.

"Fast" beds require considerable number of heat transfer panels. Bubbling beds need large cross sectional area (or stacked design).

For regeneration (calcination), low  $CO_2/O_2$  ratio is favorable, thus vigorous mixing of solids will be necessary to prevent hot-spot under high  $O_2$  partial pressure conditions.

## Research in Niigata Univ. (under preparation)



Carbonator: BFB

ID 9.3 cm, bed height 0.3 m

Gas vel. 0.22 m/s at 600 °C

Expected  $CO_2$  removal 78%

(assumed max conv. = 0.1)

Calciner: Fast bed

ID 2.2 cm, height 1.93 m

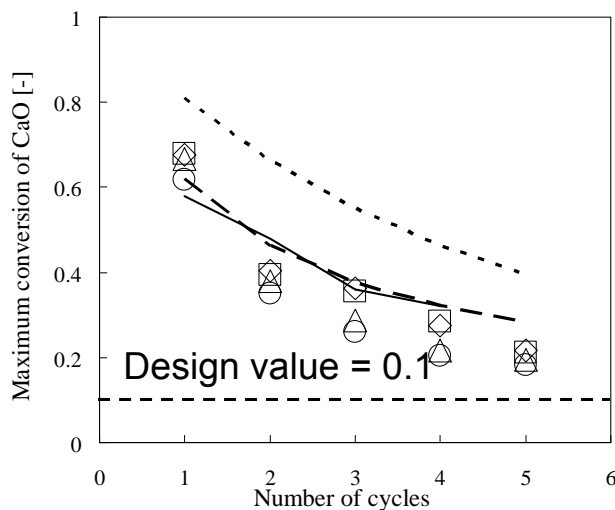
Gas vel. 6 m/s at 950 °C

$CO_2/O_2 = 1$

## Ongoing research in Niigata Univ.

### Deactivation and attrition of calcined limestone

Deactivation occurred after repeating cycles.

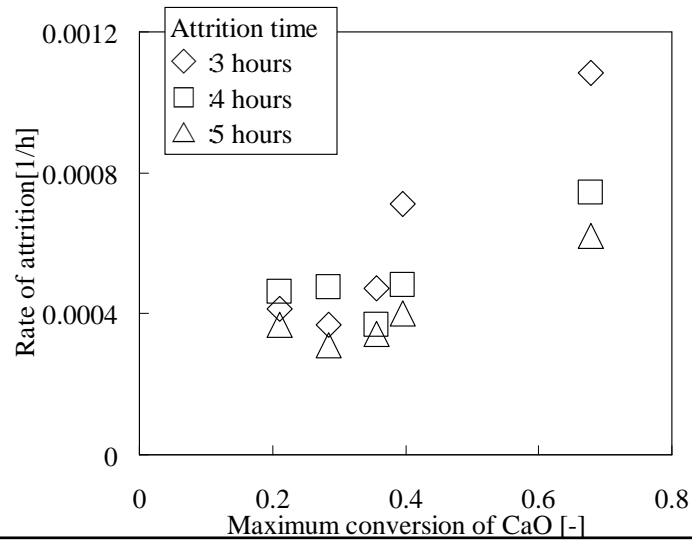


Miyairi, K.,  
Shimizu, T.,  
15<sup>th</sup> SCEJ  
Symp. on  
Fluidization  
and Particle  
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2009

Ongoing research in Niigata Univ.

## Deactivation and attrition of calcined limestone

Attrition rate decreased with deactivation.



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